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Gas chromatographic separations of all 136 tetra- to octapolychlorinated dibenzo-pdioxins and polychlorinated dibenzofurans on nine different stationary phases'

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ABSTRACT

All 49 polychlorinated dibenzo-p-dioxins and 87 polychlorinated dibenzofurans containing 4 to 8 chlorines have been synthesized and purified as individual compounds in quantitative amounts. These standards have been chromatographed on a series of nine fused-silica capillary gas chromatography (GC) columns containing silicone stationary phases of diverse polarity (100% methyl, 5% phenyl methyl, 50% phenyl methyl, 50% methyl trifluoropropyl, 50%, 75%, 90% and 100% cyanopropyl and liquid crystalline smectic). The data, expressed in a series of GC chromatograms and in tables of relative retention times, are the most comprehensive to date with regard to individual congeners and variety of stationary phases and provide a confirmation of much earlier work. The information shows that all 136 compounds, including the biologically important 2,3,7&substituted congeners, can be separated from each other mostly with two stationary phases. However, possible variation in GC conditions and stationary phases necessitates assessment of the resolution of near eluting isomers. Comparisons and contrasts to previously published reports have also been noted.

INTRODUCTION

Polychlorinated dibenzo-p-dioxins (PCDDs) and polychlorinated dibenzofurans (PCDFs) are two classes of environmental contaminants which are present as impurities in a variety of industrial chemicals such as chlorophenols and polychlorinated biphenyls (PCBs). They are also formed in heat processes particularly in the incineration of municipal waste [l], and recently have been found in the bleaching of pulp and paper [2,3]. The PCDDs (49 congeners) and PCDFs (87 congeners) of

^a This study was conducted as project 32/85 of the Working Group on Halogenated Hydrocarbon Environmental Contaminants, Food Chemistry Commission, International Union of Pure and Applied Chemistry (IUPAC).

TABLE I

^a CDD = chlorinated dibenzo-p-dioxin; CDF = chlorinated dibenzofuran; M = mono; Di = di; $Tr = tri$; $T = tetra$; $Pn = penta$; $Hx = hexa$; $Hp = hepta$; $O = octa$.

biological interest contain 4 to 8 chlorines and these along with their acronyms are shown in detail in Table I. For brevity and simplicity when describing and comparing different congeners, the chlorine substitution is usually designated without using commas to separate the arabic numbers. PCDDs and PCDFs with the 2,3,7,8 configuration (17 congeners) are the most toxic in experimental animals. These congeners have been found in wildlife [4], human tissues [5], and food samples [6], and are the most important for purposes of identification and quantification. Mass spectrometry (MS) provides unequivocal identification of the elemental composition (molecular formula) of a substance, but gas chromatographic (GC) techniques are required to separate and identify those compounds having the same elemental composition but different structure or constitution (isomers). For example, MS can detect all 22 TCDD isomers but GC is needed to specify which of these has the 2,3,7,8_substitution.

A plethora of data has been published $[7-44]$ in the last 10 years on the separation of the isomers and congeners of PCDDs/PCDFs using gas chromatography. In the late 1970s and early 1980s most of this work was performed using stationary phases coated on inert supports (packed GC columns). In the last 5 years or so virtually all separations have been carried out on the better resolving capillary columns where the stationary phase is coated on or chemically bonded to the glass surface of the wall. In particular, studies on the separation of all 38 TCDFs and 22 TCDDs on both non-polar (methyl silicone) [15,16,35] and polar (cyanopropyl silicone) $[14-16, 18, 23, 35, 40]$ columns are available and the GC properties of all 128 PCDDs/PCDFs containing 4 to 6 chlorines are known for both non-polar and polar phases [9,1 I, 13,19-22, 25-29,30, 32-34,39,42]. However, information on GC separations is often lacking or incomplete for: (a) the homologues higher than tetrachloro for medium polar (methyl-phenyl silicones) phases and (b) for all homologues on certain uncommon phases (e.g. methyl-trifluoropropyl silicones and liquid crystalline). In addition much of the data are scattered throughout the literature, and are incomplete for the congeners of biological interest. Some discrepancies also exist between separations and relative peak assignments reported by various research groups. As a result, we obtained authentic standards of all 136 PCDDs/PCDFs of biological interest in pure separate quantitative amounts (not mixtures) by synthesis and exchange. Their elution order and separations were then examined on a wide variety of stationary phases on either bonded or wall-coated fused-silica capillary columns using the liquid phases most commonly in the field as well as these advocated by Yancey [45-47] in his review of liquid phases. However, this report does not attempt to predict or provide a theoretical model for separation based on molecular structure.

EXPERIMENTAL

Definition and nomenclature

The terms, congener, isomer, and homologue are often used to describe the different kinds of PCDDs and PCDFs but unfortunately the usage of these terms has been inconsistent and confusing. In this paper we use congener to refer to any member of the class of PCDDs or PCDFs regardless of the degree of chlorination. In this respect there are 75 possible congeners of PCDDs and 135 congeners of the PCDFs (Table I). Isomers are compounds with the same elemental compositon or molecular formula (number and kinds of atoms) but with different structural or configurational arrangement. It is well known that there are 22 structural isomers of the TCDDs. Homologues usually refer to groups of compounds with a common structure $(e.g.,)$ CDD or CDF) but differing by a constant increment of common atoms (e.g. number of chlorines). Mass spectrometry (MS) can readily distinguish one homologue group of PCDDs/PCDFs from another simply by their mass to charge ratio. However, MS is generally unable to specify which isomer of any homologue is present. Such an assignment requires the technique of gas chromatography. A combination of both GC and MS techniques permits, in theory, the identification of all congeners of the PCDDs/PCDFs.

The nomenclature for the numbering of the chlorine positions in the CDD and CDF ring system is that given in Fig. 1 subscribing to the usage of Chemical Abstracts

Fig. 1. Chemical structures of the congeners of PCDDs/PCDFs.

TABLE II

SOURCES AND PURITY OF THE 49 PCDDs WITH 4 TO 8 CHLORINES

 \degree Pyrolysis: chlorophenol(s) in equal molar amounts with copper filings and potassium carbonate dissolved in freshly prepared excess methanolic KOH, evaporated to dryness, and heated at 280-300°C for 1 h as in refs. I I and 17; number in parenthesis after pyrolysis refers to different reaction mixtures of chlorophenols or routes used to obtain the same product. S1: synthesis by alkali condensation of chlorocatechol or chloroguiacol with chloronitrobenzene. S2: synthesis by alkali condensation of 2-bromo-3,4,5-trichlorophenol followed by selective crystallization of mixture. RP-HPLC: reversed-phase high-performance liquid chromatography (number in parenthesis refers to number ofcycles needed to obtain stated purity). NP-HPLC: normal-phase HPLC.

(CA). In addition, CA nomenclature states that the lowest number possible must be used when alternatives exist. In particular, in the TCDF series $(cf.$ Table I for acronyms), 1678 is used and not 2349; in the PnCDF series, 12678, 12679, 13678, and 14678 are the correct numbering and not the alternatives 23489, 13489, 23479 and 23469, respectively, and; in the HxCDF series, the isomeric designations are 124679 and 134678 and not 134689 and 234679, respectively. All of the above alternatives have been used in some instances in the literature.

Standards

PCDDs. All 49 congeners of the PCDDs of homologue from tetrachloro- to octachloro- were obtained as authentic individual standards in low milligram amounts either by synthesis [pyrolysis of 10–25 mg quantities of the appropriate chlorophenols or by two other routes (Sl and S2) as noted in Table II and refs. 11 and 171, or by exchange with or donation from other research groups. Table II gives the CA registry number, the systematic number [28] and the hex base number [48] along with the source, method of synthesis, and purity for each congener. The pyrolysis reaction products and other mixtures were purified to single compounds by repeated separation on preparative scale reversed-phase (RP) high-performance liquid chromatography (HPLC) columns containing octyldecylsilane (ODS) as stationary phase. The eluents were various combinations of methanol, ethanol, isopropanol and acetonitrile, some with the addition of small amounts (1 to 10%) of water similar to previous reports [10,14,15,22,23]. Several passes by HPLC were needed in some cases in order to obtain the purity stated in Table II. Particularly difficult congeners to separate and purify were the combinations of 1247/1248,12468/12479,124679/124689 and 123679/123689 and the congeners 12367, 12368 and 123469 all of which contained small amounts (3 to 10%) of other congeners. The isomer pair 1247/1248 was the only PCDD which could not be resolved by RP-HPLC but did separate nicely on normal-phase HPLC.

The assignment of the configuration of each PCDD was unequivocal in most cases. Some PCDDs/PCDFs have been authenticated by X-ray diffraction [49,50] and these primary standards have been compared in the literature to other congeners with regard to their physical and chemical properties. For instance, it has been established [l **1,171** by photolysis of congeners of known constitution that those congeners containing adjacent chlorines on both sides of the planar ring next to the oxygen $(e.g.,\)$ 123789-HxCDD, 12389-PnCDD and 1289-TCDD) are sterically strained and are the isomers that elute last on most CC columns. Because of the Smiles rearrangement [51,52], pyrolysis of many chlorinated phenols generated two isomers which differed only in the relative orientation of the two lateral aromatic rings. By judicious choice of the starting material, other mixtures can be synthesized containing one of a pair from a previous pyrolysis. This allowed an assignment to be made in most cases as detailed earlier [17]. In addition, in many cases certain standards were available from other research groups or commercial sources. These were sometimes synthesized by other routes (e.g. condensation of a chlorocatechol with a nitrobenzene) and could be directly compared to one synthesized by the chlorophenol route. In one case of Smiles-rearranged products, the 1237/1238-TCDD pair, specific assignment could be made. Photolysis of individual aliquots of 12367 and 12389-PnCDD caused loss of chlorine giving TCDD mixtures. Comparison of these TCDD mixtures on GC to the two isomers comprising the 1237/1238-TCDD pair from the Smiles rearrangement

(themselves separable by RP-HPLC) showed that 1237-TCDD eluted first on polar GC columns but second on RP-HPLC. Conversely, the peak from the 1237/1238- TCDD pair eluting second on polar GC columns but first on RP-HPLC was in fact the 1238-TCDD isomer. Six congener pairs synthesized as mixtures, 1246/1249-TCDD, 1247/1248-TCDD, 12468/12479-PnCDD, 12467/12489-PnCDD, 124679/124689-HxCDD, and 123679/123689-HxCDD could not be unequivocably assigned. In these cases, the isomer with the lowest numerical nomenclature designation has been assigned as the earliest eluting compound on RP-HPLC (NP-HPLC for the 1247/ 1248-TCDD pair). For example with the pair 1246/ 1249-TCDD, the first eluting isomer on RP-HPLC has been assigned arbitrarily as 1246-TCDD and this designation is used in the GC chromatograms. All standards were checked for purity using GC with electron capture (EC) detection supplemented in some cases with flame detection. Identity was confirmed by using GC with MS detection.

Purified amounts (50 to 500 μ g) of standards were weighed on a Cahn 26 electrobalance which has been calibrated for accuracy against a known mass. Precision of weighing of 200 μ g quantities was less than 2%. Usually a stock solution of the weighed material was made up in 10 ml of toluene in a volumetric flask and serial dilutions of this concentration were caried out to produce 1.0 ng/ μ l and 0.1 ng/ μ l solutions in 5 ml volumes.

PCDFs. Most of the *87* PCDFs of homologue tetrachloro- to octachloro- were synthesized either by dehydrogenation of the corresponding chlorinated diphenyl ether [53] (synthesized from the chlorinated phenol and chlorinated diphenyl iodonium salt; route S2 in Table III) or by condensation of the corresponding chlorophenol and chloronitrobenzene followed by reductive cyclization [20] (route Sl in Table III). In the few cases where a PCDF could not be synthesized from these two routes, they were prepared by two other methods (S3 or S4 in Table III) or obtained from other research groups or commercial sources. The reaction mixtures were separated by RP-HPLC in a similar fashion as for the PCDDs except that fewer passes were needed to obtain the stated purity since PCDF mixtures show greater resolution on RP-HPLC than do the PCDDs. Impurities in the PCDFs mixtures were often one homologue lower than the desired product *i.e.* dechlorination rather than dehydrochlorination. Assignment of isomeric structure of the PCDFs is given in more detail in refs. 15, 16 and 22 with the additional support in this work that many of the PCDFs were supported by NMR data [20].

Instrumentation

The gas chromatographs used were Varian models 3500 and 6000 equipped with capillary columns and on-column injectors. The ovens and injectors could be programmed independently of each other. The method of injection was by the solute focusing on-column technique whereby the injector was kept at 80°C [below the boiling point of the solvent (usually toluene)], and then the injector heat ramped quickly (1 to 3 min) to $230-260^{\circ}$ C – the final column temperature. The oven and the column itself were initially held at 120° C, just above the boiling temperature of the solvent, then heated rapidly to $160-180^{\circ}$ C and then slowly (2 to 3° C per min) to the final temperature. Carrier gas for the capillary columns was helium at a linear velocity of 30 cm/s corresponding to a volume flow of about 2.0 ml/min. Chart speed for the chromatograms varied between 2.0 and 4.0 cm/s and volume of injection between 0.6

TABLE III

SOURCES AND PURITY OF THE 87 PCDFs WITH 4 TO 8 CHLORINES

(Continued on $p.$ 140)

No.	Isomer	CA registry number	Systematic number	Base hex number	Source and purity ^a
HxCDFs					
1	123467	79060-60-9	115	fcF	S1:98%
2	123468	69698-60-8	116	faF	S2: 98%
3	123469	91538-83-9	117	f9F	S1; 98%
$\overline{4}$	123478	70648-26-9	118	f6F	(a) S2; with 123489-F; RP-HPLC, 95%
					(b) Cambridge Isotope, U.S.A., 98%
5	123479	91538-84-0	119	f5F	SI: 98%
6	123489	92341-07-6	120	f3F	S2 with 123478-F; RP-HPLC (2), 98%
7	123678	57117-44-9	121	7eF	(a) Canadian Wildlife; 98%
					(b) $S2$; 98%
8	123679	92341-06-5	122	7dF	C. Rappe; S3; 98%
9	123689	75198-38-8	123	7bF	S2; RP-HPLC; 98%
10	123789	72918-21-9	124	77F	(a) Cambridge Isotope; 98%
					(b) C. Rappe, S3; 98%
\mathbf{H}	124678	67562-40-7	125	bcF	S2: 98%
12	124679	75627-02-0	126	bdF	S2; RP-HPLC, 98%
13	124689	69698-59-5	127	bbF	S2; RP-HPLC, 98%
14	134678	71998-75-9	128	deF	S2: 98%
15	134679	92341-05-4	129	ddF	(a) Wellington; 70%
					(b) C. Rappe; S3, 98%
16	234678	60851-34-5	130	eeF	(a) $S2$; 98%
					(b) Canadian Wildlife; mixture
HpCDFs					
1	1234678	67562-39-4	131	feF	S2: 98%
$\mathbf{2}$	1234679	70648-25-8	132	fdF	S2; RP-HPLC, 98%
3	1234689	69698-58-4	133	fbF	D. Firestone; 98%
$\overline{\mathbf{4}}$	1234789	55673-89-7	134	f7F	S2; RP-HPLC; 98%
OCDF					
1	12346789	$1010 - 77 - 1$	135	ffF	(1) Analabs, U.S.A.; 98%
					(2) S ₂ ; 98%

TABLE III *(continued)*

Sl Synthesis from corresponding chlorophenol and chloronitrobenzene followed by reductive cyclization [20].

s2 Synthesis from corresponding chlorophenol and chlorodiphenyl iodonium salt to chlorodiphenyl ether followed by palladium acetate cyclization [53].

s3 Synthesis from corresponding chlorophenol and iodobenzene followed by palladium acetate cyclization [53].

s4 Synthesis by nucleophilic displacement of corresponding chlorophenol on chlorobenzene followed by palladium acetate cyclization.

RP-HPLC Reversed phase high-performance liquid chromatography (number in parenthesis is number of collection cycles needed to obtain stated purity).

and 2.0 μ . Detection was usually carried out with a nickel-63 (8 mCi) electron-capture detector kept at 300°C with a nitrogen make-up gas flow of 20 ml/min. A flame ionization detector with air and hydrogen gases was used to monitor impurities in certain cases.

GC capillary columns

Table IV lists, for each GC stationary phase, the manufacturer, coating,

dimensions and temperature programming. These polysiloxane columns are grouped as follows: (i) non-polar (DB-1; 100% methyl and DB-5; 5% phenyl); (ii) medium polar (DB-17, OV-17; 50% phenyl-methyl and DB-210; trifluoropropyl); (iii) polar (DB-225, CPS-1, SP-233 1, CP-Sil 88; all cyanopropyl) and (iv) other (SB-smectic).

The chromatograms shown in the figures for the homologues of the PCDDs/PCDFs have been derived from injection of six composite standards of about 0.2 ng/ μ l each of all isomers of a specific homologue (Table I). Since the analytes have been detected using electron-capture detection (ECD), the absolute response for a given isomer varied by as much as an order of magnitude even though equal amounts were injected. Assignment of a given GC peak to a specific isomer within a group was made by the separate injection of solutions containing one or two isomers along with a retention time (RT) standard, the earliest eluting isomer of that homologue group. These RT reference standards were: (1) for the PCDDs; 1368-TCDD, 12468-PnCDD and 124679-HxCDD and (2) for the PCDFs; 1368-TCDF, 13468-PnCDF and 123468- HxCDF. A combination of results from these injections allowed an unequivocable assignment of the elution order of a specific isomer within a composite mixture. As the retention time windows of most homologue groups of the PCDDs/PCDFs overlap to some extent on capillary columns, particularly polar phases, the injections were carried out by individual homologues. Isomers which co-eluted or eluted near each other were co-injected to define further their degree of resolution.

RESULTS

Three mixtures of the PCDDs (22 tetra, 14 penta and 10 hexa) and three mixtures of the PCDFs (38 tetra, 28 penta and 16 hexa) were prepared at a concentration of 0.2 $ng/µ$ for each congener. The GC elution pattern of these six solutions for nine different GC stationary phases on capillary columns are presented in Figs. 2-10 and their relative retention times (RRT) are listed in Tables V-XIII. The elution order, and degree of separation are best seen from the figures which represent possible separations when all isomers are present in a mixture. For those isomers which did not produce two peaks by co-injection even though they had slightly different relative retention time (RRT), the one first reported on the figure has the shorter RRT. The RRT in the nine tables are based on the earliest eluting isomer of that particular group being arbitrarily assigned a value of 1 .OOO. The RRT of the six RT reference standards are also listed in each table enabling an estimate to be made of the degree of overlap between homologues as a function of stationary phase. The absolute retention time of a selected isomer can be approximated from the tables but should be used only for guidance since this parameter will change from column to column, with usage, and GC conditions. The higher chlorinated hepta- and octa-congeners are not shown in these figures and tables since they are readily resolved from each other on all columns investigated. For all columns, the elution order of the PCDDs is 1234679 and 1234678 for the two hepta isomers followed by octa-dioxin. For the PCDFs, the corresponding elution order for the four hepta isomers is 1234678, 1234679, 1234689 and 1234789, and then octa-furan.

As mentioned in he experimental part, six pairs of PCDDs (12 congeners), separable by HPLC, could only be effectively separated on GC with the liquid crystalline smectic phase. However their exact structure and hence elution order on

GC SILICONE FUSED-SILICA CAPILLARY COLUMNS USED FOR SEPARATION OF PCDDs AND PCDFs AND THEIR TEMPERATURE
PROGRAMS PCDFs AND THEIR TEMPERATU GC SILICONE FUSED-SILICA CAPILLARY COLUMNS USED FOR SEPARATION OF PCDDs AND PROGRAM

GC (either a then b or b then a) could not be definitely assigned. Since these congeners have arbitrarily been assigned with the lowest number in nomenclature eluting first on RP-HPLC, this uncertainly is noted by an asterisk for the data from the smectic column in Fig. lob and Table XIII. The actual structure of these pairs and hence their GC and HPLC elution order could be determined either by chemical means (dechlorination or chlorination to known congeners) or by physical means (X-ray diffraction crystallography or GC-matrix isolation Fourier-transform infrared spectroscopy).

DISCUSSION

The separations shown in the figures and the RRT in the tables are those that can be obtained using the standards, GC columns, and conditions as described in the experimental section. No particular effort was made to optimize GC conditions such as injector temperature, type of injection, gas flow, temperature programming or other variables to maximize resolution of certain congeners paricularly those which are 2,3,7,8-substituted. Thus it may be possible to obtain slightly different separations than those listed by using other conditions. In this regard we have compared separations on columns purchased at different times from the same manufacturer and with different lot numbers. With both DB-5 and DB-210, identical separations and patterns were obtained from both columns. In the case of the cyanopropyl phases, CP-Sil88 and SP-233 1, columns purchased at different times showed slight differences in resolution in isolated cases $e.g.$ isomers co-eluting on one column were found to have some degree of separation on a second different column.

One phenomenon we did notice with the SP-2331 column was the change in elution order and even improved separation as the column deteriorated. After several months of use, the chromatographic peak shape degraded as evidenced by tailing peaks -a phenomenon often attributed to oxygen attack on the stationary phase. Nevertheless, a few separations now occurred which previously with a new column and better peak shape were not possible. For example, 12378-PnCDF was completely separated from the previous co-eluter 12348-PnCDF (now earlier) but co-eluted with the previously separated 12346-PnCDF. 1678-TCDF and 1234-TCDF were now separate peaks on the deteriorated column but co-eluted on the new chromatographic column. On the other hand, 12378-PnCDD and 12369-PnCDD now co-eluted on the degraded column where previously they were well separated. A similar loss in resolution with the TCDF isomers has been noted by Swerev and Ballschmiter [54] for the SP-2331 phase although no new separations were reported.

Most of the 136 PCDDs and PCDFs can be readily separated from each other using a combination of conventional GC phases. Exceptions to this are certain pairs of PCDDs containing mostly 124-substitution. These are 1247/1248, 1246/1249, 12468/12479, 12467/12489, 124679/124689 and 123679/123689 for which there is little or no resolution on the common GC phases. However, the newly developed smectic liquid crystalline phase [36,37,41] is unique in its resolving powers and is readily capable of resolving to base line the above six PCDD pairs. In fact it has not been possible until the advent of this stationary phase to specify the relative proportions of these isomer pairs. The smectic crystalline phase appears to have other unusual properties. Our separations for the TCDDs differ from those of Mahle *et al.* [41] in

Fig. 2. (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a DB-1 fused-silica bonded phase capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs, 14 PnCDDs, and 10 HxCDDs on a DB-1 fused-silica bonded phase capillary column.

Fig. 3.

Fig. 3 (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a DB-5 fused-silica bonded phase capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs, 14 PnCDDs, and 10 HxCDDs on a DB-5 fused-silica bonded phase capillary column.

Fig. 4. (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a DB-17 fused-silica bonded phase capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs, 14 PnCDDs, and IO HxCDDs on a DB-17 fused-silica bonded phase capillary column.

Fig. 5. (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a DB-210 fused-silica bonded phase capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs, 14 PnCDDs, and 10 HxCDDs on a DB-210 fused-silica bonded phase capillary column.

Fig. 6.

Fig. 6. (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a DB-225 fused-silica bonded phase capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs. 14 PnCDDs, and 10 HxCDDs on a DB-225 fused-silica bonded phase capillary column.

Fig. 7. (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a CPS-I fused-silica capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs, 14 PnCDDs, and 10 HxCDDs on a CPS-1 fused-silica capillary column.

Fig. 8.

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Fig. 8. (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a SP-233 1 fused-silica capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs, 14 PnCDDs, and 10 HxCDDs on a SP-2331 fused-silica capillary column.

Fig. 9. (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a CP-Sil 88 fused-silica capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs, 14 PnCDDs, and 10 HxCDDs on a CP-Sil 88 fused-silica capillary column.

Fig. 10. (a) GC-ECD tracing of the separation of 38 TCDFs, 28 PnCDFs, and 16 HxCDFs on a liquid crystalline smectic fused-silica capillary column. (b) GC-ECD tracing of the separation of 22 TCDDs, 14 PnCDDs, and 10 HxCDDs on a liquid crystalline smectic fused-silica capillary column; congeners marked with an asterisk are pairs which cannot be unequivocably assigned $(cf.$ text.).

TABLE V

GC RETENTION TIMES OF PCDFjPCDD ISOMERS RELATIVE TO RT REFERENCE STAN-DARDS ON A DB-1 FUSED-SILICA BONDED PHASE CAPILLARY COLUMN

TABLE V *(continued)*

a RT reference standard is 1368-TCDF.

* RT reference standard is 13468-PnCDF.

' RT reference standard is 123468-HxCDF.

^d RT reference standard is 1368-TCDD.

e RT reference standard is 12468-PnCDD.

 J RT reference standard is 124679-HxCDD.

some of the elution orders. In this respect Riehle *et al.* [37] have noticed that the retention times and even relative elution orders of the PCDDs/PCDFs on various smectic columns changed according to the thermal history of the column. This enigmatic property of the smectic phase may explain the different reported elution orders.

Thus for any sample containing mixtures of PCDDs and PCDFs, it is possible to assign the exact isomeric configuration of any peak except the six pairs noted provided two or more GC columns are used. For samples such as fly ash, waste material or PCB extracts, many congeners will be present and probably more than two GC columns would be necessary for detailed specific identification. This data would also be useful in the selection of stationary phases for use in dual or multi-dimensional GC [21,55].

TABLE VI

GC RETENTION TIMES OF PCDFjPCDD ISOMERS RELATIVE TO RT REFERENCE STAN-DARDS ON A DB-5 FUSED-SILICA BONDED PHASE CAPILLARY COLUMN

TABLE VI (continued)

Comparison to published reports

The data generated in this report is a confirmation of much previous work and agrees in all respects with that reported in the literature except for the following.

TCDDs. Taylor et al. [23] of Wright State University noted a difference between their work and two earlier reports [10,11] in the elution order of three TCDDs, 1268-, 1278- and 1279-TCDD, on both a DB-5 and SP-2331 columns and were unable to distinguish the correct elution order. Subsequently Gurka et al. [56], Gelbaum et al. [30] and Donnelly et al. [34] all investigated these elution orders and agreed with the elution order of Buser and Rappe [11]. We also find the same elution order as did Buser and Rappe [11]. Harden et al. [40] reported further separations of all 22 TCDDs and all 38 TCDFs on 4 additional GC phases including 3 in this report (DB-225, SP-2401) equivalent to DB-210, and SP-2250 equivalent to DB-17). While we are in complete agreement with their TCDF assignments, we differ markedly for those on the 1268-, 1278-, and 1279-TCDD isomers and suspect they may have used the earlier Dow assignment [10] rather than that of Buser and Rappe [11] which we and others believe to be correct.

TABLE VII

GC RETENTION TIMES OF PCDFjPCDD ISOMERS RELATIVE TO RT REFERENCE STAN-DARDS ON A DB-17 FUSED-SILICA BONDED PHASE CAPILLARY COLUMN

With regard to the isomer pair, 1246/1249-TCDD, we readily separate them on HPLC using an ODS packing and a 5% water in methanol eluent, but O'Keefe et *al.* [14], Taylor *et al.* [23] and Wagel *et al.* [57] all reported no such separation. Moreover, we are unable to separate these two isomers on the polar CP-Sil88 phase even though both Zoller and Ballschmiter [32] and Gelbaum et *al.* [30] achieved a small separation on that column. There is some degree of separation (20-30%) for this pair of TCDD isomers on the CPS-1 and DB-210 columns and a clear resolution on the liquid crystalline smectic.

The isomer pairs, 1237/1238-TCDD, from the Smiles rearrangement are not easy to separate and even more difficult to assign their structures using conventional means. We are unable to obtain significant GC separations of this pair on the first four GC phases listed in Table IV but do obtain almost 100% resolution on the other five phases (four of them being cyanopropyl and one the smectic). Because of our photolytic experiments with 12389-PnCDD and 12367-PnCDD, we assign 1237- TCDD and 1238-TCDD as the earlier and later eluters, respectively, on the cyanopropyl GC phases. On the other hand, the elution order on the smectic phase and

TABLE VIII

GC RETENTION TIMES OF PCDFjPCCD ISOMERS RELATIVE TO RT REFERENCE STAN-DARDS ON A DB-210 FUSED-SILICA BONDED PHASE CAPILLARY COLUMN

TABLE VIII *(continued)*

on RP-HPLC is reversed *i.e.* 123%TCDD first followed by 1237-TCDD. In this regard we do not agree with the separations of Brown *et al.* [44] on DB-225 for 1234-, 1237and 123%TCDD. However column differences may account for some of this incompatability. Lastly, in this report and that of Buser and Rappe [17], the elution of 1239-TCDD on OV-1 is subsequent to that of 237%TCDD whereas it was given by Oehme and Kirschmer [18] as preceding 2378-TCDD.

PnCDDs and HxCDDs. We can find no significant differences that have not already been noticed between our results and those reported previously for the above two homologues.

TCDFs. The early pioneering work by Mazer and co-workers [15,16] on this homologue and by Bell and Gara [22] on all PCDFs resulted in a comprehensive data base on their GC properties on methyl silicone (SE-54 equivalent to DB-5) and the SP-2330 phase (cyanopropyl). Mazer *et al.* [15] made two corrections to their list (ref. 15, p. 1648), one for 1236-TCDF on SE-54 and one for 2368-TCDF on SP-2330, both of which we are in agreement. Bell and Gara [22] also reported for the 1246-TCDF

TABLE IX

GC RETENTION TIMES OF PCDF/PCDD ISOMERS RELATIVE TO RT REFERENCE STAN-DARDS ON A DB-225 FUSED SILICA BONDED PHASE CAPILLARY COLUMN

isomer an RRT on DB-5 greater than that of 2378-TCDF which is different from this report and ref. 15 and probably incorrect.

For the close eluting isomer pair 1349/1267-TCDF, Waddell et *al.* [35] reported an RRT on DB-5 slightly later than 2378-TCDF whereas all other work including this one find an RRT slightly earlier than 2378-TCDF on this non-polar phase. Whether this difference is due to variation in phase type, GC conditions or other factors is not certain. All of the above groups report the same RRT for these isomers on SP-2330.

Ligon and May [21] gave separations on Silar 10C, a 100% cyanopropyl phase similar to CP-Sil 88. Those shown for 1249- and 2468-TCDF, for 1269-TCDF (after 2378-TCDF), and 2346/2367-TCDF (reversed in ref. 21) all differ significantly from other work including the present one on the equivalent phase, CP-Si188. Separations reported by all groups for the other homologues of the PCDFs are the same for the two types of cyanopropyl columns. Whether the differences for the TCDFs are due to manufacturing processes or misidentification is not clear.

Recently, the US EPA [58] reported an isomer specific separation on a DB-225 phase of 2378-TCDF from all other TCDFs including the near elutors, 2347-TCDF

TABLE X

GC RETENTION TIMES OF PCDF/PCCD ISOMERS RELATIVE TO RT REFERENCE STAN-DARDS ON A CPS-1 FUSED-SILICA CAPILLARY COLUMN

(earlier) and 1239-TCDF (later). Both this work and that of Harden *et al.* [40] find the same elution order as above but the resolution between 2378-TCDF and 1239-TCDF is small. Again this difference may be affected by injection techniques, temperature programming or even slight variation in the phases themselves.

PnCDFs. The earliest complete collection of data for this homologue is that of Rappe [19] followed by Bell and Gara [22] both on the polar phase SP-2330. For the isomers, 13479 - and 13478 -PnCDF, the elution order given by Bell and Gara $[22]$ is as written above while that of Rappe [19] is reversed. We agree with Bell and Gara [22] for both the SP-2330 and CP-Sil88 phases i.e. 13479-PnCDF elutes before 13478-PnCDF and the latter co-elutes with 12368-PnCDF. Similarly, Bell and Gara [22] found that the isomer 12678-PnCDF (equivalent to 23489-) co-eluted with 12469-PnCDF followed much later by 23468-PnCDF. Rappe [19] has the elution order 23468- and 12678-PnCDF -we concur with the 1985 data [22]. The elution order given by Rappe [19] for the above pentafuran isomers has been used for structure assignment by Zoller and Ballschmiter [32] in chlorophenol pyrolysates, by Abraham *et al.* [59] in a rat study, by Wakimoto *et al.* [60] in determination of PCDFs in PCBs, and by Oehme *et*

TABLE XI

GC RETENTION TIMES OF PCDF/PCCD ISOMERS RELATIVE TO RT REPRENCE STAN-DARDS ON A SP-2331 FUSED-SILICA CAPILLARY COLUMN

al. [61] in PCDFs from magnesium smelting. Interestingly, Rappe et al. [62] correctly report the elution order of 12678-/23468-PnCDF in their more recent study on urban air.

HxCDFs. The only discrepancy we noted for the HxCDF homologues is for the 134678/124678-HxCDF pair which we and Bell and Gara [22] report as resolved peaks in the above order on the polar SP-2330 and CP-Sil 88 columns and not as first reported [19] in reverse order, and subsequently used to assign isomers in fly ash [60] and urban air [62].

2378-substituted PCDDIPCDF. Since the 2378-substituted PCDD/PCDF congeners are so important in both environmental and biological samples, their separation for the tetra-, penta- and hexa-homologues is shown in more detail in Tables XIV and XV for all nine GC phases. From Table XIV for the PCDDs, it is seen that the polar cyanopropyl columns such as SP-2331 and CP-Si188, and the smectic liquid crystalline phase can separate both 2378-TCDD from the other 21 isomers and the other four higher chlorinated 2378~substituted PCDDs from their respective isomers. The medium polar DB-17 is next best in the separation of these compounds with only

TABLE XII

GC RETENTION TIMES OF PCDFjPCCD ISOMERS RELATIVE TO RT REFERENCE STAN-DARDS ON A CP-Sil88 FUSED-SILICA CAPILLARY COLUMN

1469-TCDD interfering. The non-polar DB-1 and DB-5 columns have many congeners which co-elute with the 2378-substituted PCDDs.

In the case of the 2378-substiuted PCDFs (Table XV), the situation is more complicated. Only the DB-17 column was successful in our hands in separating 2378-TCDF from the other 37 TCDFs. The more polar DB-210, SP-2331 and CP-Sil 88 all had difficulty separating 2348-TCDF from 2378-TCDF. Interestingly, the other bonded phase methyl-phenyl (50%) (OV-17) column did not separate 1469- from 2378-TCDF. There may be major differences in the resolving power of this and other phases depending on batch number and manufacturer in addition to GC conditions such as temperature and injection mode. However the OV-17 column did allow complete isomeric identification of 12378-PnCDF which DB- 17 did not (interference from 12346-PnCDF on the latter column). DB-210 was also successful in the separation of 12378- from 12348-PnCDF which the cyanopropyl and methyl silicone columns could not accomplish. The non-polar DB- 1 and DB-5 have many isomers in the tetra- and penta-series which potentially overlap with the 2378-substituted PCDFs. As mentioned previously, the DB-225 column we used could not separate 1239-TCDF

TABLE XIII

GC RETENTION TIMES OF PCDFjPCDD ISOMERS RELATIVE TO RT REFERENCE STAN-DARDS ON A LIQUID CRYSTALLINE SMECTIC FUSED-SILICA CAPILLARY COLUMN

^{*a*} Congeners marked with an asterisk are pairs which cannot be unequivocably assigned $(cf. text)$.

from 2378-TCDF although this has been reported by others [58]. With regard to the HxCDFs, Table XV shows that a combination of using both a non-polar and medium or high polar column will allow unequivocal identification of all 2378-substituted HxCDFs and indeed of all 16 isomers.

The data show that all the isomers and congeners can be separated from each other by a combination of a minimum of two columns. In particular, all of the biologically important 2378-substituted PCDDs/PCDFs commonly found in animal species are readily differentiated on two GC columns. It is emphasized that, due to factors such as variation in columns and GC conditions, each laboratory must test the resolving power of their GC columns by demonstrating that those isomers eluting in the immediate vicinity of the congener in question as judged from this report are in fact suitably resolved. This information is the most comprehensive to date with regard to the number of single congeners reported and with respect to the number and range of polarity of the capillary columns. It also provides a confirmation of much earlier work. Except for six closely related PCDD pairs, the isomeric PCDD-PCDF content can be

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COLUMNS COLUMNS

ISOMERS WHICH COULD POTENTIALLY INTERFERE WITH THE 2378-SUBSTITUTED PCDDs ON FUSED-SILICA SILICONE CAPILLARY GC

TABLE XIV

TABLE XIV

TABLE XV TABLE XV

ISOMERS WHICH COULD POTENTIALLY INTERFERE WITH THE 2378-SUBSTITUTED PCDFs ON FUSED-SILICA SILICONE CAPILLARY GC ISOMERS WHICH COULD POTENTIALLY INTERFERE WITH THE 2378-SUBSTITUTED PCDFs ON FUSED-SILICA SILICONE CAPILLARY GC **COLUMNS** COLUMNS

Percent resolution between isomers in brackets; GC conditions as in Table IV. Percent resolution between isomers in brackets; GC conditions as in Table IV.

^a Resolution for SP-2331 in brackets shown first for a new column and then for one which has deteriorated with use. 4 Resolution for SP-2331 in brackets shown first for a new column and then for one which has deteriorated with use. unambiguously specified for complex environmental samples containing many peaks or for the simpler biologically incurred samples containing mostly 2378-PCDD/ PCDFs.

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